

**TABLE 3.** Names and idealized chemical compositions of selected minerals produced by biological mineralization processes (expanded from Dove and Weiner 2003, Table 1)

Name	Formula
<b>Elements</b>	
α-sulfur	S
γ-sulfur (Rosickyite)	S
<b>Carbonates</b>	
Amorphous calcium carbonates (at least 5 forms)	CaCO <sub>3</sub> ·H <sub>2</sub> O or CaCO <sub>3</sub>
*Aragonite	CaCO <sub>3</sub>
*Calcite	CaCO <sub>3</sub>
Hydrocerussite	Pb <sub>3</sub> (CO <sub>3</sub> ) <sub>2</sub> (OH) <sub>2</sub>
Lansfordite	MgCO <sub>3</sub> ·5H <sub>2</sub> O
Magnesite	MgCO <sub>3</sub>
Mg-calcite	(Mg <sub>x</sub> Ca <sub>1-x</sub> )CO <sub>3</sub>
*Monohydrocalcite	CaCO <sub>3</sub> ·H <sub>2</sub> O
Nesquehonite	Mg(CO <sub>3</sub> )·3H <sub>2</sub> O
Protodolomite	CaMg(CO <sub>3</sub> ) <sub>2</sub>
Rhodochrosite	MnCO <sub>3</sub>
Siderite	FeCO <sub>3</sub>
*Vaterite	CaCO <sub>3</sub>
<b>Phosphates</b>	
Amorphous calcium phosphate	variable (at least 6 forms)
Amorphous calcium pyrophosphate	Ca <sub>2</sub> P <sub>2</sub> O <sub>7</sub> ·2H <sub>2</sub> O
Bakhchisaraitsevite	Na <sub>2</sub> Mg <sub>5</sub> (PO <sub>4</sub> ) <sub>4</sub> ·7H <sub>2</sub> O
*Brushite	Ca[PO <sub>3</sub> (OH)]·2H <sub>2</sub> O
*Carbonate-hydroxylapatite	Ca <sub>5</sub> (PO <sub>4</sub> ) <sub>3</sub> (OH)
Chlorapatite	Ca <sub>5</sub> (PO <sub>4</sub> ) <sub>3</sub> Cl
*Fluorapatite	Ca <sub>5</sub> (PO <sub>4</sub> ) <sub>3</sub> F
*Hannayite	Mg <sub>3</sub> (NH <sub>4</sub> ) <sub>2</sub> H <sub>4</sub> (PO <sub>4</sub> ) <sub>4</sub> ·8H <sub>2</sub> O
*Hydroxylapatite	Ca <sub>5</sub> (PO <sub>4</sub> ) <sub>3</sub> (OH)
*Monetite	Ca[PO <sub>3</sub> (OH)]
*Newberryite	Mg[PO <sub>3</sub> (OH)]·3H <sub>2</sub> O
Octacalcium phosphate	Ca <sub>8</sub> H <sub>2</sub> (PO <sub>4</sub> ) <sub>6</sub>
*Struvite	Mg(NH <sub>4</sub> )(PO <sub>4</sub> )·6H <sub>2</sub> O
Hazenite	KNaMg <sub>2</sub> (PO <sub>4</sub> ) <sub>2</sub> ·14H <sub>2</sub> O
Vivianite	Fe <sub>3</sub> <sup>2+</sup> (PO <sub>4</sub> ) <sub>2</sub> ·8H <sub>2</sub> O
*Whitlockite	Ca <sub>18</sub> H <sub>2</sub> (Mg,Fe) <sub>2</sub> (PO <sub>4</sub> ) <sub>14</sub>
<b>Sulfates</b>	
Aphthitalite	K <sub>3</sub> Na(SO <sub>4</sub> ) <sub>2</sub>
Ardealite	Ca <sub>2</sub> [PO <sub>3</sub> (OH)](SO <sub>4</sub> )·4H <sub>2</sub> O
Barite	BaSO <sub>4</sub>
Celestine	SrSO <sub>4</sub>
*Epsomite	MgSO <sub>4</sub> ·7H <sub>2</sub> O
*Gypsum	CaSO <sub>4</sub> ·2H <sub>2</sub> O
*Hexahydrate	MgSO <sub>4</sub> ·6H <sub>2</sub> O
Jarosite	KFe <sub>3</sub> <sup>3+</sup> (SO <sub>4</sub> ) <sub>2</sub> (OH) <sub>6</sub>
Melanterite	Fe <sup>2+</sup> SO <sub>4</sub> ·7H <sub>2</sub> O
Schwertmannite	Fe <sub>16</sub> O <sub>16</sub> (OH) <sub>9,6</sub> (SO <sub>4</sub> ) <sub>3,2</sub> ·10H <sub>2</sub> O
<b>Nitrates</b>	
Gwihaibaite	(NH <sub>4</sub> )NO <sub>3</sub>
<b>Sulfides</b>	
Acanthite	Ag <sub>2</sub> S
Amorphous pyrrhotite	Fe <sub>1-x</sub> S (x = 0 - 0.17)
Galena	PbS
Greigite	Fe <sub>3</sub> S <sub>4</sub>
Hydrotroilite	Fe <sub>5</sub> NH <sub>2</sub> O
Mackinawite	(Fe,Ni) <sub>9</sub> S <sub>8</sub>
Marcasite	FeS <sub>2</sub>
Pyrite	FeS <sub>2</sub>
Pyrrhotite	Fe <sub>7</sub> S <sub>8</sub>
Orpiment	As <sub>2</sub> S <sub>3</sub>
Sphalerite	ZnS
Wurtzite	ZnS

\* Minerals found in humans (after Yoder 2002).

**TABLE 3.—CONTINUED**

Name	Formula
<b>Oxides</b>	
Amorphous iron titanate	Fe <sup>2+</sup> TiO <sub>3</sub>
Amorphous iron oxide	Fe <sub>2</sub> O <sub>3</sub>
Amorphous manganese oxide	Mn <sub>2</sub> O <sub>4</sub>
*Anatase	TiO <sub>2</sub>
Ilmenite	FeTiO <sub>3</sub>
*Maghemite	Fe <sub>2,67</sub> O <sub>4</sub>
*Magnetite	Fe <sub>3</sub> O <sub>4</sub>
*Periclase	MgO
Ice	H <sub>2</sub> O
<b>Hydroxides and hydrous oxides</b>	
Birnessite	NaMn <sub>4</sub> O <sub>8</sub> ·3H <sub>2</sub> O
*Brucite	Mg(OH) <sub>2</sub>
Ferrihydrite	Fe <sub>4-5</sub> (OH) <sub>12</sub>
Goethite	α-FeO(OH)
Lepidocrocite	γ-FeO(OH)
Todorokite	NaMn <sub>6</sub> O <sub>12</sub> ·3-4H <sub>2</sub> O
<b>Chlorides</b>	
Atacamite	Cu <sub>2</sub> Cl(OH) <sub>3</sub>
*Halite	NaCl
*Sylvite	KCl
<b>Fluorides</b>	
Fluorite	CaF <sub>2</sub>
Hieratite	K <sub>2</sub> SiF <sub>6</sub>
<b>Silicates</b>	
Amorphous silica	SiO <sub>2</sub> ·nH <sub>2</sub> O
*Quartz	SiO <sub>2</sub>
<b>Organic crystals*</b>	
Abelsonite	Ni <sup>2+</sup> C <sub>31</sub> H <sub>32</sub> N <sub>4</sub>
Ca malate	C <sub>4</sub> H <sub>4</sub> CaO <sub>5</sub>
Ca tartrate	C <sub>4</sub> H <sub>4</sub> CaO <sub>6</sub>
Carpathite	C <sub>24</sub> H <sub>12</sub>
Earlandite	Ca <sub>3</sub> (C <sub>6</sub> H <sub>5</sub> O <sub>7</sub> ) <sub>2</sub> ·4H <sub>2</sub> O
Glushinskite	MgC <sub>2</sub> O <sub>4</sub> ·2H <sub>2</sub> O
Guanine	C <sub>5</sub> H <sub>3</sub> (NH <sub>2</sub> )N <sub>4</sub> O
Hartite	C <sub>20</sub> H <sub>34</sub>
Hoganite	Cu(CH <sub>3</sub> COO) <sub>2</sub> ·H <sub>2</sub> O
Ildrialite	C <sub>22</sub> H <sub>14</sub>
Kratochvilite	C <sub>13</sub> H <sub>10</sub>
Lindbergite	MnC <sub>2</sub> O <sub>4</sub> ·2H <sub>2</sub> O
Moolooite	Cu <sub>2</sub> O <sub>4</sub> ·nH <sub>2</sub> O
Paceite	CaCu(CH <sub>3</sub> COO) <sub>2</sub> ·6H <sub>2</sub> O
<b>Paraffin hydrocarbon</b>	
Sodium urate	C <sub>5</sub> H <sub>3</sub> N <sub>4</sub> NaO <sub>3</sub>
*Urea	CO(NH <sub>2</sub> ) <sub>2</sub>
Uricite	C <sub>5</sub> H <sub>4</sub> N <sub>4</sub> O <sub>3</sub>
*Weddelite	CaC <sub>2</sub> O <sub>4</sub> ·2H <sub>2</sub> O
*Whewellite	CaC <sub>2</sub> O <sub>4</sub> ·H <sub>2</sub> O

\* Minerals found in humans (after Yoder 2002).